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questions Name _____

Common-Ion Effect 1)

The pH of a solution that contains 0.818 M acetic acid ($K_a = 1.77 \times 10^{-5}$) and 0.172 M sodium acetate is ____.

A) 4.077 B) 5.434 C) 8.571 D) 8.370 E) 9.922

2) Calculate the pH of a solution prepared by

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dissolving 0.37 mol of formic acid (HCO_2H) and 0.23 mol of sodium formate ...

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