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chemistry. □ Describe oxidation and

reduction (redox) reactions. □ Discuss

the different forms of matter: elements,

compounds, and mixtures. □ Explain

the difference between solutions,

suspensions, and emulsions. □ Explain

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pH and the pH scale.

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organic chemistry. The study of substances that contain the element carbon. oxidation. A chemical reaction that combines a substance with oxygen to produce an oxide. oxidation-reduction. Also known as redox; a chemical reaction in which the oxidizing agent is reduced and the reducing agent is oxidized. oxidizing agent.

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STUDY. PLAY. Chemistry. the science that deals with the composition, structures, and properties of matter and how matter changes under different conditions. Organic

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Chemistry. the study of substances that contain the element carbon.

Inorganic chemistry.

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Chapter 12 Basics of Chemistry. This activity will help you to learn key terms in cosmetology chemistry. STUDY. PLAY. acidic solution. A solution that has a pH below 7.0 (neutral). alkaline solution. A solution that has a pH above 7.0 (neutral). alkanolamines.

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A chemical reaction that combines a substance with oxygen to produce an oxide. oxidation-reduction. (aka redox) a chemical reaction in which the oxidizing agent is reduced (by losing oxygen) and the reducing agent is oxidized (by gaining oxygen). oxidizing agent. Substance that releases oxygen.

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A stable physical mixture of 2 or more substances. ALKANOLAMINES. Alkaline substances used to neutralize acids or raise the PH of many hair

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products. EMULSION. An unstable physical mixture of 2 or more immiscible substances, plus a special ingredient called a "emulsifier".

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This book covers the basic concepts found in introductory high-school and college chemistry courses.

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Provides questions and answers similar to what is found on state licensing exams in cosmetology.

Organic Chemistry: Structure, Mechanism, Synthesis, Second Edition, provides basic principles of this fascinating and challenging science, which lies at the interface of physical and biological sciences. Offering accessible language and engaging examples and illustrations, this valuable introduction for the in-depth chemistry course engages students and gives future and new scientists a new approach to understanding, rather than merely memorizing the key concepts underpinning this fundamental area. The book builds in a logical way from chemical bonding to resulting molecular structures, to the

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corresponding physical, chemical and biological properties of those molecules. The book explores how molecular structure determines reaction mechanisms, from the smallest to the largest molecules—which in turn determine strategies for organic synthesis. The book then describes the synthetic principles which extend to every aspect of synthesis, from drug design to the methods cells employ to synthesize the molecules of which they are made. These relationships form a continuous narrative throughout the book, in which principles logically evolve from one to the next, from the simplest to the most complex examples, with abundant connections between the theory and applications. Featuring in-book solutions and instructor PowerPoint slides, this

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