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1. increasing the surface area.
 2. increasing the temperature.
 3. increasing the strength of the intermolecular forces.
- 1 and 2 only.

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required to melt one mole of a solid is called the: heat of fusion.

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2p C Carbon atoms do exhibit this bonding pattern in very rare circumstances, but in most cases, they form four bonds and have no lone pairs. Methane, CH₄, is a typical

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1 . The instantaneous rate is the rate of a reaction at any particular point in time, a period of time that is so short that the concentrations of reacta

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phenomenon of attracting and
retaining the molecules of a substance
by a solid (or a liquid) on its surface

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resulting into a higher concentration of the molecules on the surface is known as adsorption. 2.

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