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Chemistry Chapter 7 Test. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. christina_lombardi. Terms in this set (20) A _____ is a shorthand way of describing a chemical reaction. chemical equation. When two chemicals are mixed and the beaker gets warm, it usually indicates that a chemical reaction has taken place.

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7 Chemical Formulas and Chemical Compounds

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7.3 Ozone: Pollutant and Protector 7.4 Chlorofluorocarbons: A Chemical Success Story Gone Wrong Review Skills The presentation of information in this chapter assumes that you can already perform the tasks listed below. You can test your readiness to proceed by answering the Review Questions at the end of the chapter.

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This video explains the answers to the practice quiz on Chapter 7 - 8, which can be found here: <https://goo.gl/yxsQH4>

Chapter 7 - 8 Practice Quiz - YouTube

Chemistry Chapter 7 Test Multiple Choice Identify the choice

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that best completes the statement or answers the question.

____ 1. A chemical formula includes the symbols of the elements in the compound and subscripts that indicate a. atomic mass of each element. b. number of atoms or ions of each element that are combined in the compound.

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STANDARDIZED TEST PREP 1. b 2. d CHAPTER 7 □
HOMEWORK 7-2 VOCABULARY 1. d 2. c 3. e 4. a 5. g 6. h
7. b 8. f GRAPHIC ORGANIZER Name of Compound or Ion
Formula of Compound Copper(II) ion Cu^{2+} Ammonium ion
 NH_4^+ Bismuth(III) bromide BiBr_3 Stannous fluoride SnF_2
Magnesium fluoride MgF_2 Iron(III) cyanide $\text{Fe}(\text{CN})_3$ Sulfide
ion S^{2-} Potassium ...

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