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Esterification Reaction Mechanism - Carboxylic Acid  
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NCEA Level 3 Organic Chemistry) Synthesis of an ester  
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Organic Molecules: Alcohols, Carboxylic Acids and Esters  
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Arora Synthesis Of Ethyl benzoate (sweet smelling ester)  
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The preparation and the practical report will be completed,

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in lesson time, under supervision. The preparation will be conducted in pairs, but individual reports will be written.

Stage 2 Chemistry Practical (S): PREPARATION OF AN ESTER. The preparation and the practical

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Table 1 –Reagents for preparation of esters Add 4 drops of concentrated sulphuric acid to each test tube. Pour about

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150 ml of tap water in a 250 ml beaker. Place the test tubes in the water and heat the water on a hot plate to a temperature of about  $60^{\circ}\text{C}$  –  $75^{\circ}$  Leave the test tubes in the hot water bath for 15 minutes.

## ~~Preparation and identification of Esters—Physical ...~~

Part A: Preparation of Ester. Measure 15 mL of the alcohol provided and 10 mL glacial (pure) acetic acid into the 50 mL pear-shaped Quickfit flask. Slowly add about 1 mL concentrated sulfuric acid and a few boiling chips. Assemble the glassware for reflux. Reflux the mixture for 30 minutes.

Part B: Isolation of Pure Ester

~~STAGE 2 CHEMISTRY~~

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Practical Investigation: Preparation of an Ester. Response 1. Skills checklist (Teacher record) Part A: Teacher observations: Performing reflux Clamped correctly. Vapours reached the top of the condenser. Use of separating funnel Careful. Performing distillation Clamped correctly. Adaptor joint under strain. Thermometer too low.

### ~~Preparation of Ester: Student Report~~

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Esters can be synthesized artificially in labs by combining alcohols and acids of different strengths. This lab contained the synthesis of an ester by allowing different carboxylic acids to react with alcohols, using sulphuric acid as a catalyst, to create esters and allow the student to identify them by their distinct smell.

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Practical Assessment 5 (b) The preparation of an organic liquid compound (e.g. the preparation of an ester). It is also expected that candidates will be aware of the procedures involved in the purification and re-distillation at an appropriate temperature of the product obtained. The Preparation of Ethyl Ethanoate. Reaction

~~A-Level Chemistry - Home~~

To make a small ester like ethyl ethanoate, you can gently heat a mixture of ethanoic acid and ethanol in the presence of concentrated sulphuric acid, and distil off the ester as soon as it is formed. This prevents the reverse reaction

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happening. It works well because the ester has the lowest boiling point of anything present.

~~preparation of esters — chemguide~~

5. EXEMPLARS OF PRACTICAL WORK AS FORMAL ASSESSMENT TASKS TERM 1: PRACTICAL WORK  
KNOWLEDGE AREA: MATTER AND MATERIALS 5.1  
PREPARATION OF ESTERS AND SMELL IDENTIFICATION  
INTRODUCTION Esters have a very fruity smell. Naturally occurring esters are found in fruits. Esters can be synthesised by the reaction of a carboxylic acid and an alcohol.

~~PHYSICAL SCIENCES~~

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assessment tasks 338 PREPARATION FILE - Grade 12 Doc

Scientia Method: 1. Number the test tubes from 1 – 6 and place it in the test tube rack. 2. Add the correct amount of the substances to the test tubes as indicated in the table. Test tube Carboxylic acid Alcohol 1 1 cm<sup>3</sup> ethanoic acid 1 cm<sup>3</sup>

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methanol 2 1 cm<sup>3</sup> ethanoic acid 1 cm<sup>3</sup> ethanol

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