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1. How many valence
electrons does a neutral

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magnesium (Mg) atom have? 2. What is the charge on a magnesium ion? What does magnesium have to do to form such an ion, and why does it tend to do so? 3. How many valence electrons does a single neutral chlorine (Cl)

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atom have? 4. Answers

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and Sublevels ...

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question 5, how many
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(Ave1 is a ha AC5a) 5th 7.

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in a p sublevel ...

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Chemistry 9.2 Answers
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Section 9.2 1. For each of the following pairs of elements, underline the one that would replace the other element in a compound.

a. calcium, tin e. iron, copper
b. bromine, fluorine f. iodine, chlorine c.

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aluminum, potassium g.
silver, lead d. zinc, sodium
2.

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and Change 0 Chapter 8 .

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1.7, the CF bond is polar
covalent. (b) $EN_{Cl} - EN_P = 3.0 - 2.1 = 0.9$. Since $0.5 < EN_{Cl} - EN_P < 1.7$, the PCl bond is polar
covalent. (c) $EN_{Cl} - EN_{Li}$

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Section 8.5 Chemistry Answers

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