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Types of Chemical Reactions Introduction to Balancing Chemical Equations Worksheet 6 2 Chemical Reaction Worksheet 6-2 Chemical Reaction Types Name Period

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Glencoe pp. 277-279, 284-291 Identify each chemical reaction as a synthesis (combination), decomposition, single-replacement, double-replacement, or combustion reaction.

1. $2\text{SO}_2 + \text{O}_2 \rightarrow 2\text{SO}_3$ 2. $\text{Al}_2(\text{SO}_4)_3 + 3\text{Ca}(\text{OH})_2 \rightarrow 2\text{Al}(\text{OH})_3 + 3\text{CaSO}_4$ 3. $2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$ 4.

Worksheet 6-2 Name Chemical Reaction Types ... - Winters ...
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1. Write the chemical equations and balance each of the following word equations.

a) Aluminum metal reacts with iron (II) oxide powder to produce aluminum oxide solid and iron metal.

$$2\text{Al}(s) + 3\text{FeO}(s) \rightarrow \text{Al}_2\text{O}_3(s) + 3\text{Fe}(s)$$

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Worksheet 6.2 Word Equations

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Chemical Reactions WS6-2 Word Equations.doc 3 Answers to Worksheet 6.1 Writing and Balancing Equations 1. Write the chemical equations and balance each of the following word equations. a) Aluminum metal reacts with iron (II) oxide powder to produce aluminum oxide solid and

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Worksheet 6-2 Chemical Reaction Types Name Period

Identify each chemical reaction as a synthesis (combination), decomposition, single- replacement, double-replacement, or

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combustion reaction. 1, 2, 3. 4. 5. 6, 8. 9. 10. $2\text{SO}_2 + \text{O}_2 + +$
 $3\text{CaSO}_4 + 5\text{O}_2 + 2\text{H}_2\text{O}$ $\text{Mg} + 2\text{AgNO}_3 + 2\text{Ag} + 2\text{H}_3\text{PO}_4 +$
 6HNO_3 $\text{MgCl}_2 + 3\text{O}_2$ $2\text{Be} + \text{O}_2$ 2BeO $2\text{Al} + + 3\text{Cu}$ 2PbO_2
 $2\text{PbO} + \text{O}_2$

Ms. Chab's Website - Home

Section Quick Check Date CHAPTER 6 Section 2: Chemical
Reactions Class After reading the section in your textbook,
respond to each statement. 1. State the term for the amount
of energy that is needed for a chemical reaction to occur. 2.
Summarize the relationship between an enzyme and a
substrate. enc 3. <https://www.svcasd.org/cms/lib07/NY01913388/Centricity/Domain/245/biochem%20review.pdf>.

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Chapter 6 Section 2 Chemical Reactions Answer Key

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Six Types Of Chemical Reaction Worksheet Answers [PDF]

This is the chemical reaction where an oxygen compound and carbon compound combine together to become H_2O and CO_2 . It is the reaction where mostly an organic compound like oxygen burns yielding to water, carbon dioxide, or some other product. The combination of any substance with oxygen results in combustion.

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